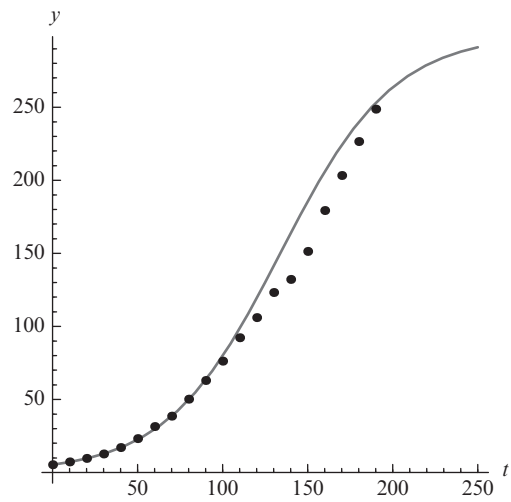


FIGURE 3.1 A population with limited resources cannot grow exponentially forever.



**FIGURE 3.2** With these parameter values, the upper limit on the population of the United States is approximately 300 million.

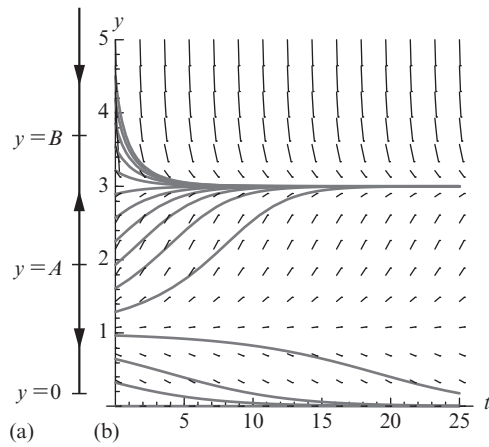


FIGURE 3.3 (a) Phase line for  $\frac{dy}{dt} = -r(1 - (1/A)y)(1 - (1/B)y)$ . (b) Several solutions together with the slope field for  $\frac{dy}{dt} = -0.25(1 - y)(1 - y/3)y$ . The phase portrait shows that all nontrivial solutions other than if  $y_0 = 1, y(t) = 1$ . If  $y_0 < 1$ , then  $\lim_{t \rightarrow \infty} y(t) = 0$ . If  $y_0 > 1$ , then  $\lim_{t \rightarrow \infty} y(t) = 3$ .

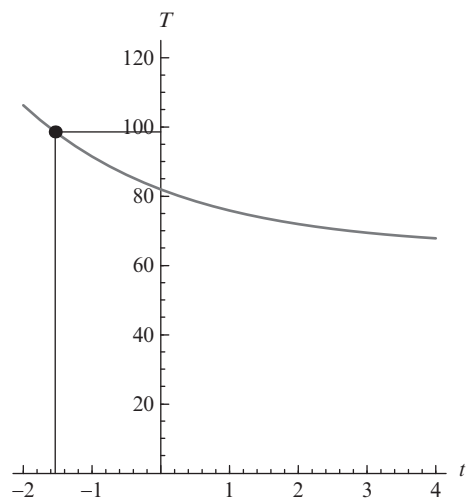


FIGURE 3.4 Graph of  $T(t) = 17(7/17)^{t/2} + 65$ .

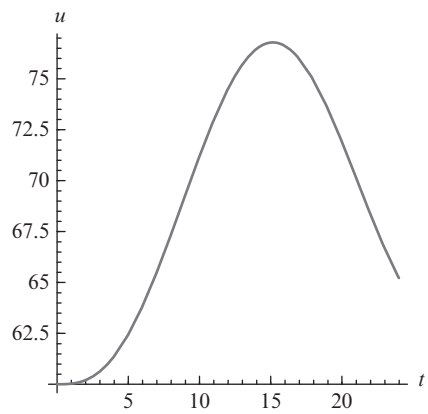


FIGURE 3.5 Modeling the temperature in a building over the course of a 24-h day.

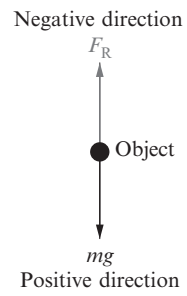


FIGURE 3.6 Force diagram.

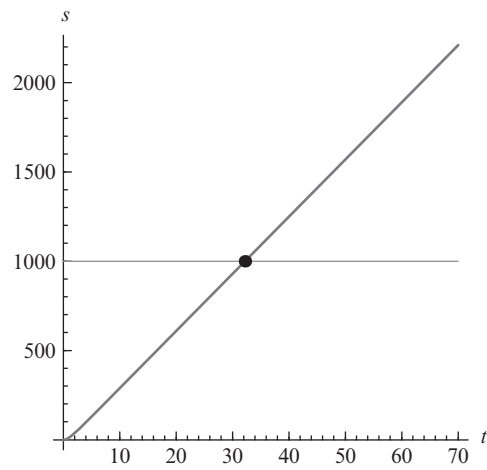
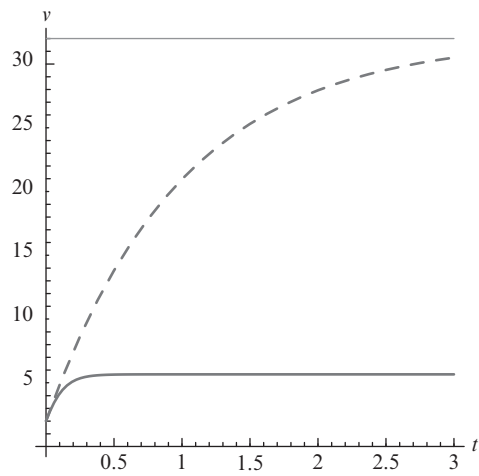


FIGURE 3.7 Graph of  $s(t) = 32t + 30e^{-t} - 30$ .



**FIGURE 3.8** The velocity functions from Example ?? (dashed) and Example ?. Notice how the different forces due to air resistance affect the velocity of the object.



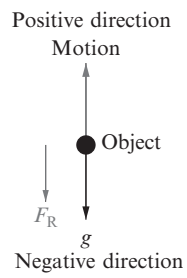


FIGURE 3.9 By drawing a force diagram, we see that  $g$  and  $F_R$  are in the negative direction.

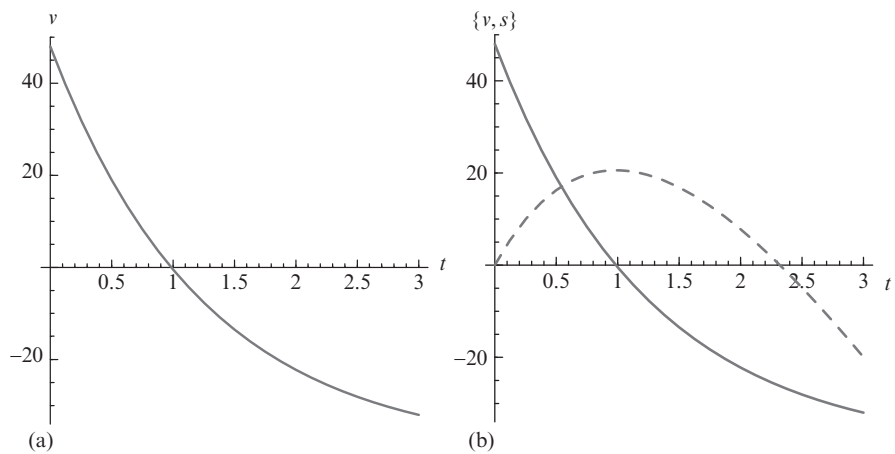


FIGURE 3.10 (a) Graph of  $v(t) = 88e^{-4t/5} - 40$ . (b) Graph of  $v(t)$  together with  $s(t) = 110 - 40t - 110e^{-4t/5}$  (dashed).

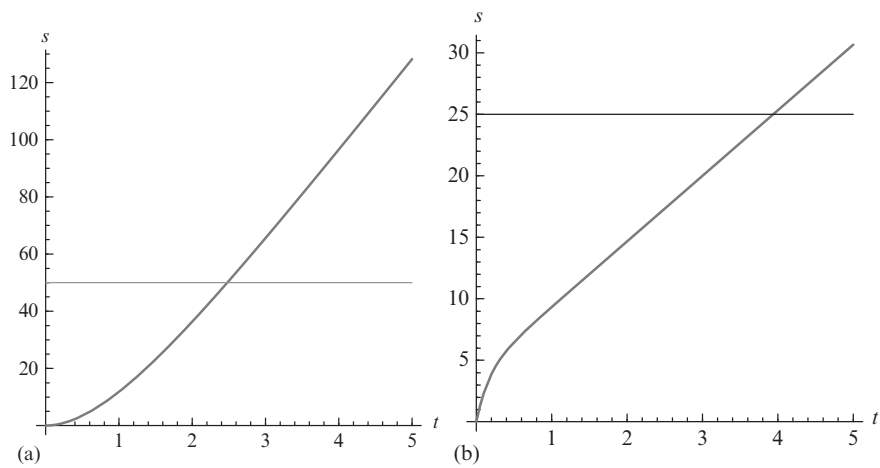
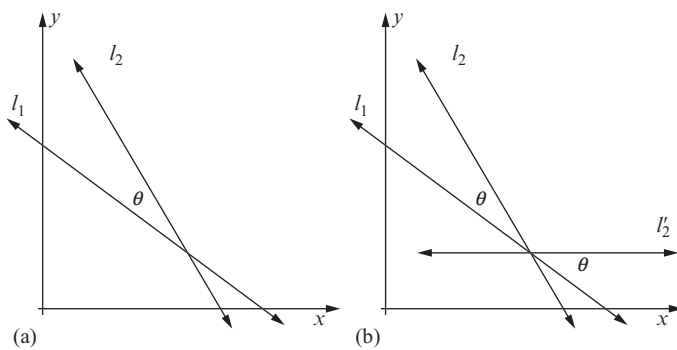


FIGURE 3.11 (a) Graph of  $s(t) = 32e^{-t} + 32t - 32$ . (b) Graph of  $s(t) = 3.99722 - 3.9972e^{-6t} + \frac{16}{3}t$ .



**FIGURE 3.12** (a) Calculating the angle of intersection of two non-parallel lines. (b) Drawing the parallel can help compute the angle.

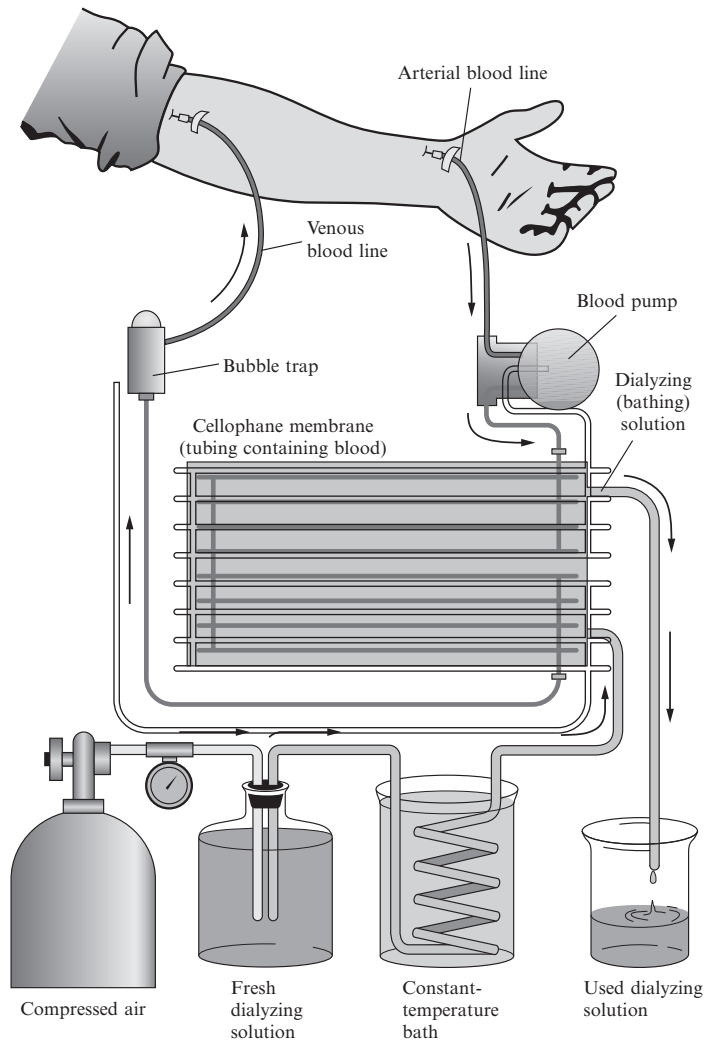


FIGURE 3.13 Diagram of a kidney dialysis machine.

**TABLE 3.1** Half-Lives of Various Nuclides

<b>Element</b>	<b>Nuclide</b>	<b>Half-Life</b>	<b>Element</b>	<b>Nuclide</b>	<b>Half-Life</b>
Aluminum	$^{26}\text{Al}$	$7.4 \times 10^5$ years	Polonium	$^{209}\text{Po}$	100 years
Beryllium	$^{10}\text{Be}$	$1.51 \times 10^6$ years	Polonium	$^{210}\text{Po}$	138 days
Carbon	$^{14}\text{C}$	5730 years	Radon	$^{222}\text{Rn}$	3.82 days
Chlorine	$^{36}\text{Cl}$	$3.01 \times 10^5$ years	Radium	$^{226}\text{Ra}$	1700 years
Iodine	$^{131}\text{I}$	8.05 days	Thorium	$^{230}\text{Th}$	75,000 years
Potassium	$^{40}\text{K}$	$1.2 \times 10^9$ years	Uranium	$^{238}\text{U}$	$4.51 \times 10^9$ years

**TABLE 3.2** U.S. Population and Values of  $y(t)$ 

<b>Year (<math>t</math>)</b>	<b>Actual Population (in millions)</b>	<b>Value of <math>y(t) =</math> <math>5.3 e^{0.03t}</math></b>	<b>Year (<math>t</math>)</b>	<b>Actual Population (in millions)</b>	<b>Value of <math>y(t) =</math> <math>5.3 e^{0.03t}</math></b>
1800 (0)	5.30	5.30	1870 (70)	38.56	43.28
1810 (10)	7.24	7.15	1880 (80)	50.19	58.42
1820 (20)	9.64	9.66	1890 (90)	62.98	78.86
1830 (30)	12.68	13.04	1900 (100)	76.21	106.45
1840 (40)	17.06	17.60	1910 (110)	92.23	143.70
1850 (50)	23.19	23.75	1920 (120)	106.02	193.97
1860 (60)	31.44	32.06	1930 (130)	123.20	261.83

**TABLE 3.3** U.S. Population and Values of  $y(t)$

<b>Year (<math>t</math>)</b>	<b>Actual Population (in millions)</b>	<b>Value of <math>y(t)</math></b>	<b>Year (<math>t</math>)</b>	<b>Actual Population (in millions)</b>	<b>Value of <math>y(t)</math></b>
1800 (0)	5.30	5.30	1900 (100)	76.21	79.61
1810 (10)	7.24	7.11	1910 (110)	92.23	98.33
1820 (20)	9.64	9.52	1920 (120)	106.02	119.08
1830 (30)	12.68	12.71	1930 (130)	123.20	141.14
1840 (40)	17.06	16.90	1940 (140)	132.16	163.59
1850 (50)	23.19	22.38	1950 (150)	151.33	185.45
1860 (60)	31.44	29.44	1960 (160)	179.32	205.82
1870 (70)	38.56	38.42	1970 (170)	203.30	224.05
1880 (80)	50.19	49.63	1980 (180)	226.54	239.78
1890 (90)	62.98	63.33	1990 (190)	248.71	252.94



**TABLE 3.4** Units Useful in Solving Problems Associated with Newton's Second Law of Motion

	<b>English</b>	<b>International</b>
Mass	Slug ( $1\text{b s}^2/\text{ft.}$ )	Kilogram (kg)
Force	Pound (lb)	Newton ( $\text{m kg/s}^2$ )
Distance	Foot (ft.)	Meter (m)
Time	Second (s)	Second (s)

<b>Painting</b>	<b><math>^{210}\text{Po}</math> Concentration (dpm/g of Pb)</b>	<b><math>^{226}\text{Ra}</math> Concentration (dpm/g of Pb)</b>	<b><math>\frac{1-\text{Ra}}{\text{Po}}</math></b>
<i>Washing of Feet</i>	12.6	0.26	0.98
<i>Woman Reading Music</i>	10.3		0.97
		0.30	
<i>Woman Playing Mandolin</i>	8.2	0.17	0.98
<i>Woman Drinking</i>	8.3	0.1	0.99
<i>Disciples of Emmaus</i>		0.8	0.91
<i>Boy Smoking</i>	4.8	0.31	0.94
<i>Lace Maker</i>	1.5	1.4	0.07
<i>Laughing Girl</i>	5.2	6.0	-0.15